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Study number	94229

FINAL REPORT

Algae Growth Inhibition Study of 13F-OLE with *Pseudokirchneriella subcapitata*

September 18, 2007

Kurume Laboratory
Chemicals Evaluation and Research Institute, Japan

STATEMENT

Kurume Laboratory
Chemicals Evaluation and
Research Institute, Japan

Sponsor DAIKIN INDUSTRIES, LTD.

Title Algae Growth Inhibition Study of 13F-OLE with *Pseudokirchneriella subcapitata*

Study number 94229

I, the undersigned, hereby declare that this report provides a correct English translation of the Final Report (Study No. 94229, issued on September 18, 2007).

The Study Director was changed from Masanori Seki to Mika Ono, because Masanori Seki had been reshuffled.

Date October 27, 2009

Study Director

GLP STATEMENT

Kurume Laboratory
Chemicals Evaluation and
Research Institute, Japan

Sponsor DAIKIN INDUSTRIES, LTD.

Title Algae Growth Inhibition Study of 13F-OLE with *Pseudokirchneriella subcapitata*

Study number 94229

The study described in this report was conducted in compliance with the following GLP principles:

- (1) "Standard Concerning Testing Facility Relating to New Chemical Substances" (November 21, 2003; No. 1121003, Pharmaceutical and Food Safety Bureau, Ministry of Health, Labour and Welfare; November 17, 2003, No. 3, Manufacturing Industries Bureau, Ministry of Economy, Trade and Industry; No. 031121004, Environmental Policy Bureau, Ministry of the Environment)
- (2) "OECD Principles of Good Laboratory Practice (November 26, 1997, ENV/MC/CHEM (98)17)"

This final report reflects the raw data accurately and it has been confirmed that the test data are valid.

Date September 18, 2007

Study Director Signed in original

QUALITY ASSURANCE STATEMENT

Kurume Laboratory
Chemicals Evaluation and
Research Institute, Japan

Sponsor DAIKIN INDUSTRIES, LTD.

Title Algae Growth Inhibition Study of 13F-OLE with *Pseudokirchneriella subcapitata*

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I assure that the final report accurately describes the test methods and procedures, and that the reported results accurately reflect the raw data of the study.

The inspections and audit of this study were carried out and the results were reported to the Study Director and the Test Facility Management by Quality Assurance Unit as follows.

Item of inspection / audit	Date of inspection / audit	Date of report to Study Director and Test Facility Management
Study plan draft	August 24, 2007	August 24, 2007
Study plan	August 24, 2007	August 24, 2007
Amendment of study plan	September 14, 2007	September 18, 2007
Measurement of solubility	August 25, 2007	August 29, 2007
	August 27, 2007	August 29, 2007
Start of the exposure and after the exposure	August 25, 2007	August 30, 2007
	August 27, 2007	August 30, 2007
	August 30, 2007	August 30, 2007
Raw data and final report draft	September 15, 2007	September 18, 2007
Final report	September 18, 2007	September 18, 2007

Date

September 18, 2007

Quality Assurance Unit, Head

Signed in original

CONTENTS

	Page
Title	1
Sponsor	1
Test facility	1
Objective	1
Test method	1
Applied GLP	2
Dates	2
Storage of test item, raw data, etc	2
Personnel	3
Approval of final report	3
SUMMARY	4
1. Test item	6
2. Test sample	7
3. Test materials and methods	8
4. Results and discussion	14
5. Factors that affected reliability of test results	16

Tables and figures

Table 1	pH of test solutions at start and end of exposure
Table 2	Culture temperature and light intensity in incubator
Table 3	Value of cell concentration at each time
Table 4	Growth inhibition rates at exposure level
Table 5	EC50 and NOEC on growth rate
Table 6	Result of statistical analysis
Table 7	Validity of test
Figure 1	Growth curve in each test level
Appendix 1	Composition of medium
Appendix 2	Analytical method and measured concentration of test item
Appendix 3	Calibration curve and chromatogram
Appendix 4	Solubility of test item in medium
Additional data	Results of preliminary studies

Title	Algae Growth Inhibition Study of 13F-OLE with <i>Pseudokirchneriella subcapitata</i>
Sponsor	DAIKIN INDUSTRIES, LTD. 1-1, Nishi-Hitotsuya, Settsu, Osaka 566-8585, Japan
Test facility	Kurume Laboratory Chemicals Evaluation and Research Institute, Japan 3-2-7, Miyanojin, Kurume-shi, Fukuoka 839-0801, Japan
Objective	The objective of this study is to determine the effect of 13F-OLE on growth of algae.
Test method	<p>The study was performed according to the following test methods and guidance document.</p> <ol style="list-style-type: none"> (1) Algal Growth Inhibition Test stipulated in the "Testing Methods for New Chemical Substances" (November 21, 2003; No. 1121002, Pharmaceutical and Food Safety Bureau, Ministry of Health, Labour and Welfare; November 13, 2003, No. 2, Manufacturing Industries Bureau, Ministry of Economy, Trade and Industry; No. 031121002, Environmental Policy Bureau, Ministry of the Environment, Partial amendment November 20, 2006) (2) OECD Guidelines for Testing of Chemicals, Section 2: Effects on Biotic Systems, "Freshwater Alga and Cyanobacteria, Growth Inhibition Test (Guideline 201, 23 March 2006)" (3) OECD Guidance Document 23 "Guidance Document on Aquatic Toxicity Testing of Difficult Substances and Mixtures" (September 2000)
Applied GLP	<p>The study was conducted in compliance with the following GLP principles:</p> <ol style="list-style-type: none"> (1) "Standard Concerning Testing Facility Relating to New Chemical Substances" (November 21, 2003; No. 1121003, Pharmaceutical and Food Safety Bureau, Ministry of Health, Labour and Welfare; November 17, 2003, No. 3, Manufacturing Industries Bureau, Ministry of Economy, Trade and Industry; No. 031121004, Environmental Policy Bureau, Ministry of the Environment) (2) "OECD Principles of Good Laboratory Practice" (November 26, 1997, ENV/MC/CHEM(98)17).

Dates

Study initiation date	August 24, 2007
Experimental starting date	August 25, 2007
Solubility study starting date	August 25, 2007
Bioassay starting date	August 27, 2007
Experimental completion date	August 30, 2007
Solubility study completion date	August 28, 2007
Bioassay completion date	August 30, 2007
Study completion date	September 18, 2007

Storage of test item, raw data, etc.

(1) Test item

The test sample^{*1} will be sealed in a storage vessel and stored in archives in this laboratory for ten years after the receipt of notice specified under Clause 1 or Clause 2 in Article 4, Clause 2 or Clause 3 or Clause 8 in Article 4-2, and Clause 2 in Article 5-4 or Clause 2 in Article 24 or Clause 2 in Article 25-3 of "Law Concerning Examination and Regulation of Manufacture, etc. of Chemical Substances". If it is not stable for the storage period, it is stored as long while it is kept stable. Treatment of the sample after the storage period will be discussed with sponsor.

*1 It will be stored as the common sample for storage of these studies (Study Nos. 94229, 94230 and 94231).

(2) Raw data and materials

Raw data, the study plan, documents concerning the study presented by the sponsor, the final report and necessary materials will be stored in archives in this laboratory for ten years after the receipt of the notice specified under Clause 1 or Clause 2 in Article 4, Clause 2 or Clause 3 or Clause 8 in Article 4-2, and Clause 2 in Article 5-4 or Clause 2 in Article 24 or Clause 2 in Article 25-3 of "Law Concerning Examination and Regulation of Manufacture, etc. of Chemical Substances". Treatment of raw data and materials, etc. after the storage period will be discussed with sponsor.

Personnel**Study Director :****Section 4 (Eco-toxicity test area)****Study personal****Biology :****Analysis :****Approval of final report****Study Director****Date****September 18, 2007****Signature****Signed in original**

SUMMARY

Title

Algae Growth Inhibition Study of 13F-OLE with *Pseudokirchneriella subcapitata*

Test condition

(1) Test item	13F-OLE
(2) Test organism	<i>Pseudokirchneriella subcapitata</i>
(3) Exposure duration	72 hours
(4) Test concentration	Saturated solution of test item (nominal concentration: approximately 100 mg/L) and control
(5) Type of test	Incubation with shaking (approximately 100 rpm)
(6) Preparation of test solution	The test sample and medium were mixed to prepare approximately 100 mg/L (nominal concentration), and they were stirred under closed system for approximately 48 hours. After settlement for 1 hour, test solution was prepared by taking out from the middle layer.
(7) Replicate	Six replicates / test level
(8) Volume of test solution	600 mL / test level (100 mL / test vessel) (two additional vessels for analytical chemistry of the test item were set.)
(9) Temperature in incubator	21 to 24°C, not varied more than $\pm 2^{\circ}\text{C}$
(10) Light condition	Continuous illumination using a fluorescent light [The measured light intensity was 60 to 120 $\mu\text{E}/\text{m}^2/\text{s}$ (not varied more than 20%) at the level of the test solutions during exposure period.]
(11) Measurement of cell growth	Cell concentration
(12) Analysis of concentration of test item in test solution	GC-MS analysis (at the start of the exposure, 24 and 48 hours after the start of the exposure, and the end of the exposure)

Results

(1) Solubility of test item in medium (23±1°C)	0.177 mg/L
(2) Concentration of test item in test solution	
At the start of the exposure	0.0712 mg/L
24 hours after the start of the exposure	< 0.0200 mg/L
48 hours after the start of the exposure	< 0.0200 mg/L
At the end of the exposure	< 0.0200 mg/L
(3) EC ₅₀ (E _r C ₅₀)	> 0.0139 mg/L
(4) NOEC (Growth rate 0-3d)	≥ 0.0139 mg/L

[The above-mentioned concentrations (3) and (4) are based on geometric mean of measured concentrations.]

Conclusion

This study was conducted as a limit test in order to confirm the effect of the test item on the test organisms at around the solubility of the test item in medium. The concentration of test item in the test solution at the preparation was a low value (0.0712 mg/L) compared with the solubility in medium (average: 0.177 mg/L, variation: 0.151 to 0.223 mg/L). However, considering the results of solubility test (average: 0.101 mg/L, variation: 0.0987 to 0.104 mg/L) in the fish acute toxicity study of 13F-OLE (study No. 94231), it was judged that the concentration of test item at the preparation was around solubility in medium. The concentrations of test item in the test solutions were significantly decreased during the exposure. The reason was estimated to the limit in algae growth inhibition study of the volatile substance (volatilization to headspace in the test vessel). It was concluded that the test item did not have an acute effect on the test organisms at around the solubility of the test item in medium because no adverse effect was observed in the conditions of the definitive study.

1. Test item

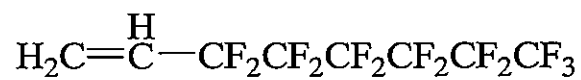
In this report, 13F-OLE has the following name etc.

1.1 Chemical name^{*2}

3,3,4,4,5,5,6,6,7,7,8,8,8-tridecafluoro-octa-1-ene

1.2 Chemical structure etc.^{*2}

Structural formula



Molecular formula	C ₈ H ₃ F ₁₃
Molecular weight	346.09
CAS Number	25291-17-2

^{*2} Information supplied by the sponsor

2. Test sample

2.1 Supplier and lot number^{*2}

Supplier	DAIKIN INDUSTRIES, LTD.
Lot number	061122HM

2.2 Purity^{*2}

Test item	99.8%
Impurity	Unknown constituent component 0.2%

2.3 Confirmation of test item

It was confirmed that infrared (IR) spectrum of the test item provided by the sponsor coincided with IR spectrum analyzed in this laboratory.

2.4 Physicochemical properties^{*2}

Appearance at normal temperature	Colorless and clear liquid	
Boiling point	106°C (760 mmHg)	
Density	1.560 g/cm ³ (20°C)	
Solubility	Water	Insoluble
	Dimethylsulfoxide	Insoluble
	Acetone	Soluble (fully miscible)

^{*2} Information supplied by the sponsor

2.5 Storage condition and confirmation of stability at storage condition

Storage condition	Dark storage place at room temperature
Confirmation of stability	The stability of the test item during the test period was confirmed by no alteration in the IR spectra of the test item before the experimental start and after the experimental completion.

3. Test materials and methods

3.1 Test organism

(1) Species

Pseudokirchneriella subcapitata (ATCC 22662)

(The old scientific name "*Selenastrum capricornutum*")

(2) Reason for selection of species

Species recommended in the test guideline

(3) Source

Pseudokirchneriella subcapitata which originally came from the American Type Culture Collection (12301 Parklawn Drive Rockville, Maryland 20852-1776 U.S.A.) on Jun. 30, 1995 and has been cultured in this laboratory was used. An algae growth inhibition test of potassium dichromate (Reagent grade, Wako Pure Chemical Industries, Ltd.) with the test organisms was conducted to confirm the reproducibility of the test system (on July 23 - July 26, 2007). The $E_rC_{50}(0-3d)$ of potassium dichromate was 1.04 mg/L. This value was within the normal range of the reference substance in this laboratory (mean \pm S.D.: 0.735 to 1.05 mg/L) [mean \pm S.D.: 0.893 \pm 0.079 mg/L (n=4)].

3.2 Culture medium

The medium recommended in OECD test guideline (Guideline 201, 23 March 2006). The composition of medium is shown in Appendix 1. Medium was used under sterile condition.

3.3 Test apparatus and equipment

(1) Test apparatus

Test vessel : Sterilized 500 mL Erlenmeyer flask (closed vessel)

(2) Test equipment

Incubator : Incubator with temperature control, continuous shaking and continuous illumination, maintained the uniform light intensity (Incubator with rotary shaker and artificial illumination, USI CO., LTD)

3.4 Test conditions

(1) Conditions of exposure

(a) Type of test

The test organisms were exposed to the test solution containing the test item added to medium. The test vessels that contain test solution and test organisms were shaken (approximately 100 rpm) during the exposure.

(b) Exposure duration

72 hours

(c) Test concentration

Based on the results of the preliminary studies, it was expected that the test solution at around the solubility in medium had no impact on the growth of algae. Therefore, the definitive study was conducted as the limit test with the saturated test solution which was stirred approximately 48 hours of test item (nominal concentration: approximately 100 mg/L). The results of preliminary studies are shown in Additional data.

(d) Control

The medium without the test item, which was treated in the same manner as the test solution (except the collection of the solution from the middle layer) was used as the control.

(e) Replicates

Six replicates / test level

(f) Initial cell concentration

The pre-culture, incubated under the same conditions as the test for 3 days and exponentially growing was used as inoculum to prepare the initial cell concentration of approximately 5×10^3 cells/mL.

(g) Operation

All operations were carried out under sterile conditions.

(h) Volume of test solution

600 mL / test level (100 mL / test vessel: two additional vessels for analytical chemistry of the test item were set.)

(2) Conditions of test environment

(a) Temperature in the incubator

21 to 24°C, not varied more than $\pm 2^\circ\text{C}$

(b) Light

Continuous illumination provided with 60 to 120 $\mu\text{E}/\text{m}^2/\text{s}$ (fluctuation range: mean $\pm 20\%$) at the level of the test solutions, using a fluorescent light with wavelength range of 400 to 700 nm.

3.5 Preparation of test solution

No correction with purity was done for the preparation of the test concentration. The test sample was employed in terms of volume using the density [$1.560 \text{ g}/\text{cm}^3$ (20°C)] for the preparation of test solution.

The test sample and medium were mixed to prepare approximately 100 mg/L (nominal concentration). Then, they were stirred by magnetic stirrer for approximately 48 hours under the closed system with little headspace. After the mixture was left for approximately 1 hour at rest, the saturated solution of test item was collected from the middle layer for use as the test solution. The test solution was divided into each test vessel.

3.6 Observation and measurements

(1) Cell growth, etc.

Biomass was shown as cell concentration.

Cell concentration was counted with particle counter (Model COULTER Z1, Beckman Coulter) at 24, 48, and 72 hours after the start of the exposure. The blank value correction was conducted by measuring simultaneously the blank values of the blank solutions without algae which were separately prepared in each test level when the test solutions were prepared. Furthermore, the cell condition for one vessel in each test level was observed under microscope (Model BX41, Olympus Co., Ltd.) at the end of the exposure.

(2) Appearance of test solution

The appearance of the test solutions was observed at the start and the end of the exposure.

(3) Water quality and environmental conditions

The pH of the test solution was measured at the start and end of the exposure. For the measurement of pH, another solution sampled from the vessel for preparation was used at the start of the exposure and one test vessel in each test level was used for the measurement at the end of exposure. The culture temperature and light intensity in the incubator were measured once a day during the exposure. The pH measurements were carried out on a portable pH meter (Model HM-21P, DKK-TOA Co.). The temperature was measured on a calibrated thermometer of glass stick type. Light intensity was measured on quantum scalar laboratory irradiance meter (Model LI-250A, LI-COR).

(4) Concentration of test item in the test solution

It was considered that the measurement was necessary with time from the results of the preliminary study, therefore, the concentration of test item in the test solution was measured at the start of the exposure, 24 and 48 hours after the start of the exposure, and the end of the exposure. At the start, the solution sampled from the container for preparation in each test level was used for the measurement. At 24 and 48 hours after the start of the exposure, required test solution sampled from the test vessel for analytical chemistry. At the end of the exposure, equal volume of the test solution was taken out from the test vessels in each test level and mixed. The concentration of the test item was analyzed by gas chromatograph-mass spectrometry (GC-MS). Analytical method and measured concentration of test item are shown in Appendix 2.

(5) Solubility of test item in medium

Solubility of the test item in medium was measured in this study because the solubility was estimated to be below 100 mg/L. Detail of the measurement and results are shown in Appendix 4.

3.7 Treatment of results

The results of the study were estimated by geometric mean of measured concentration.

(1) Calculation of concentration-inhibition rates

The mean value of biomass for each test level was plotted against time to produce growth curves. Using this curve, inhibition rates were calculated comparing with control values on growth rate.

Comparison of growth rates

The specific growth rate for a specific period was calculated as the logarithmic increase in biomass according to the following formula:

$$\mu_{i-j} = \frac{\ln N_j - \ln N_i}{t_j - t_i}$$

where

μ_{i-j} : specific growth rate from time i to j (normally day⁻¹)

N_i : measured number of cells/mL at t_i , nominal number at start t_0

N_j : measured number of cells/mL at t_j

t_i : time of ith measurement after beginning of test (day)

t_j : time of jth measurement after beginning of test (day)

Specific growth rate over the exposure duration (0-72 h) was calculated for determination of EC₅₀^{*3}. In control, specific growth rates for section-by-section were calculated for check of validity of the test.

The percentage inhibition of the cell growth at each exposure level (I_μ) was calculated as the difference between the average specific growth rate at control level (μ_c) and that at exposure level (μ_t) as:

$$I_\mu = \frac{\mu_c - \mu_t}{\mu_c} \times 100$$

*3 EC₅₀ (Median Effective Concentration) is the concentration of the test item that results in 50% reduction in growth of the test organisms during the exposure.

(2) Estimation of EC₅₀

The EC₅₀ was estimated as "> the concentration of the test item" since no less than 50% of inhibition rate was not obtained within the exposure levels. The EC₅₀ was denoted as E_rC₅₀ based on growth rate.

(3) Estimation of No Observed Effect Concentration (NOEC^{*4})

F-test was done to determine the homogeneity of variance for the data, and a significant difference was found at 5% significant level. Then, Student's *t*-test was conducted. NOEC was determined by the results of statistical analysis and cell condition.

*4 NOEC (No Observed Effect Concentration) is the highest concentration of the test item that does not cause any observed adverse effects on growth of the test organisms during the exposure.

3.8 Validity of the test

- (1) The cell growth in the control cultures should have increased by a factor of at least 16 within the 72-hour exposure period.
- (2) The mean coefficient of variation for section-by-section specific growth rates in the control cultures must not exceed 35%.
- (3) The coefficient of variation of average specific growth rates in replicate control cultures must not exceed 7%.

3.9 Treatment of numerical values

Values were rounded off in accordance with JIS Z 8401 rule B, 1999.
(JIS ; Japanese Industrial Standards)

4. Results and discussion

The contrast table of nominal concentration and geometric mean of the measured concentrations is shown below.

Nominal concentration (mg/L)	Geometric mean of measured concentration (mg/L)
100	0.0139

The value of geometric mean of the measured concentration is used in this report.

4.1 Observation of test solution and measurement of water quality variables

(1) Appearance of test solution

In the 0.0139 mg/L exposure level and control, test solutions were colorless and clear at the start of the exposure and the appearance of the test solutions at the end of the exposure were green due to the algae growth.

(2) Water quality and environmental conditions

The measured values of pH in the test solutions were 7.7 and 7.9 at the start and 10.1 at the end of the exposure. The fluctuation of pH in control was out of the range of the regulation of the test method (not increase by more than 1.5 units in ordinary practice). Temperature in incubator ranged from 22.9 to 23.9°C and light intensity was 91 to 96 $\mu\text{E}/\text{m}^2/\text{s}$. The measured values of pH of test solution are shown in Table 1, and temperature and light intensity in the incubator are shown in Table 2.

(3) Concentration of test item in test solution

The measured concentrations of the test item in the test solution were 0.0712 mg/L at the start of the exposure, and those were < 0.0200 mg/L (below the determination limit) at 24, 48 hours after the start of the exposure and the end of the exposure, respectively. The results of the measured concentrations of the test item are shown in Appendix 2.

4.2 EC_{50}

EC_{50} (E_tC_{50}) of the 13F-OLE based on the growth rate was > 0.0139 mg/L. Values of cell concentration at each time, growth inhibition rates at exposure level, and the EC_{50} are shown in Table 3, Table 4, and Table 5, respectively.

4.3 Growth curves in each test level, cell observations and NOEC

In the 0.0139 mg/L exposure level, the algae growth was close to the control. The following results of cell observation were based on the comparison with the control. No abnormality was observed in control. In the 0.0139 mg/L exposure level, the condition of cells was the same as the control.

By the results of statistical analysis about growth rate, there was no significant difference in 0.0139 mg/L exposure level compared with control.

By the results in statistical analysis and cell observation showed above, NOEC based on growth rate was ≥ 0.0139 mg/L. NOEC, the result of statistical analysis of significant difference, and growth curve are shown in Table 5, Table 6 and Figure 1, respectively.

4.4 Validity of test

Detailed result on validity of the test is shown in Table 7.

(1) Growth of control

The cell in the control grew exponentially during the exposure. At the end of exposure, it increased to 63 or more times of the number of initial cells in the control. This meets the validity of the test: the cell growth in control should have increased by a factor of at least 16 times at 72 hours after the start of the exposure.

(2) The mean coefficient of variation for section-by-section specific growth rates in the controls

The mean coefficient of variation in the control was 17.4%. It meets the validity of the test: the mean coefficient of variation in the control must not exceed 35%.

(3) The coefficient of variation of average specific growth rates in replicate controls

The coefficients of variations were 2.56% in the control. They meet the validity of the test: the mean coefficient of variation in controls must not exceed 7%.

4.5 Discussion

This study was conducted as a limit test in order to confirm the effect of the test item on the test organisms at around the solubility of the test item in medium. The concentration of test item in the test solution at the preparation was a low value (0.0712 mg/L) compared with the solubility in medium (average: 0.177 mg/L, variation: 0.151 to 0.223 mg/L). However, considering the results of solubility test (average: 0.101 mg/L, variation: 0.0987 to 0.104 mg/L) in the fish acute toxicity study of 13F-OLE (study No. 94231), it was judged that the concentration of test item at the preparation was around solubility in medium. The concentrations of test item in the test solutions were significantly decreased during the exposure. The reason was estimated to the limit in algae growth inhibition study of the volatile substance (volatilization to headspace in the test vessel). It was concluded that the test item did not have an acute effect on the test organisms at around the solubility of the test item in medium because no adverse effect was observed under the conditions of the definitive study.

With regard to the environmental conditions of the test, increase in the unit of pH in control, which is out of the range of the regulation of the test method, was observed. It was decided that the increase of the unit of pH in control was due to the limitation of the algae growth inhibition study with volatile substance (gas exchange between outside and inside of the test vessel is impossible because of the test vessel of closed system). The environmental conditions except pH were within the suitable range. Therefore, it is concluded that this study complied with the applied test guidelines.

5. Factors that affected reliability of test results

There were no factors which might have affected the reliability of the test.

Table 1 pH of test solutions at start and end of exposure

Measured concentration ^{*5} (mg/L)	pH	
	At the start	At the end
Control	7.9	10.1
0.0139	7.7	10.1

*5 Geometric mean of measured concentrations (also expressed as measured concentration in the following table)

Table 2 Culture temperature and light intensity in incubator

Time	At the start	1-day	2-day	At the end
Culture temperature (°C)	23.8	23.9	23.9	22.9
Light intensity ($\mu\text{E}/\text{m}^2/\text{s}$)	96	94	92	91

Table 3 Value of cell concentration at each time

Measured concentration (mg/L)	No.	Cell concentration ($\times 10^4$ cells/mL)			
		0 hour ^{*6}	24 hours	48 hours	72 hours
Control	1	0.499	2.67	11.9	39.4
	2	0.499	2.34	11.8	31.5
	3	0.499	2.41	11.8	41.4
	4	0.499	2.66	12.9	42.9
	5	0.499	2.38	12.2	37.8
	6	0.499	2.49	11.9	41.1
	Mean	0.499	2.49	12.1	39.0
	S.D.	0	0.144	0.409	4.07
0.0139	1	0.499	2.68	12.8	40.3
	2	0.499	2.64	13.7	34.5
	3	0.499	2.33	12.5	33.0
	4	0.499	2.54	12.0	31.0
	5	0.499	2.34	13.6	33.4
	6	0.499	2.35	12.2	38.4
	Mean	0.499	2.48	12.8	35.1
	S.D.	0	0.158	0.698	3.53

*6 The value based on the measured value of pre-culture

Table 4 Growth inhibition rates at exposure level

Measured concentration (mg/L)	No.	Growth rate (0-3d)	Inhibition rate (%)
Control	1	1.46	-
	2	1.38	-
	3	1.47	-
	4	1.48	-
	5	1.44	-
	6	1.47	-
	Mean	1.45	-
0.0139	1	1.46	-0.838
	2	1.41	2.68
	3	1.40	3.70
	4	1.38	5.18
	5	1.40	3.46
	6	1.45	0.227
	Mean	1.42	2.40

Table 5 EC₅₀ and NOEC on growth rate

Endpoint	EC ₅₀ (mg/L)	NOEC (mg/L)
Growth rate	> 0.0139	≥ 0.0139

Table 6 Result of statistical analysis

Measured concentration (mg/L)	Endpoint
	Growth rate
0.0139	n.s.
Statistical procedure	<i>F</i> -test Student <i>t</i> -test

n.s.: no significant difference

Table 7 Validity of test

< Variation for section-by-section specific growth rates in the controls >

Control No.	Mean	Standard deviation	Coefficient of variation (%)	17.4 (Mean)
1	1.46	0.24	16.7	
2	1.38	0.35	25.3	
3	1.47	0.19	13.0	
4	1.48	0.25	16.8	
5	1.44	0.27	18.8	
6	1.47	0.20	13.6	

< Variation of average specific growth rates in replicate controls >

	0-3day
Mean	1.45
Standard deviation	0.04
Coefficient of variation (%)	2.56

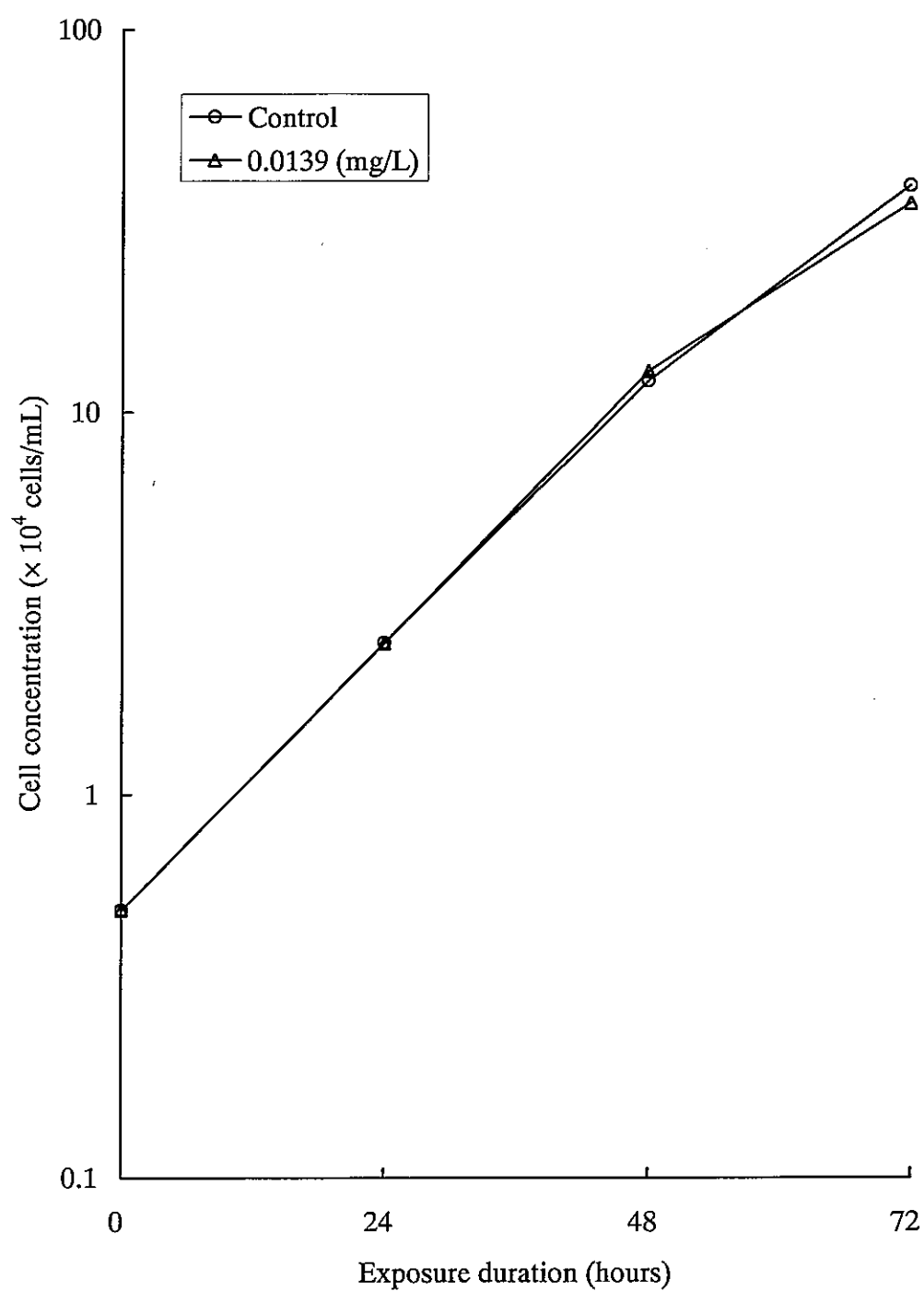


Figure 1 Growth curve in each test level.

Appendix 1

Composition of medium

Composition of OECD medium
[Guideline 201 (March 23, 2006)]

Nutrient salts	Amount	
H_3BO_3	0.185	mg
$\text{MnCl}_2 \cdot 4\text{H}_2\text{O}$	0.415	mg
ZnCl_2	0.003	mg
$\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$	0.064	mg
$\text{Na}_2\text{EDTA} \cdot 2\text{H}_2\text{O}$	0.1	mg
$\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$	0.0015	mg
$\text{Na}_2\text{MoO}_4 \cdot 2\text{H}_2\text{O}$	0.007	mg
$\text{CuCl}_2 \cdot 2\text{H}_2\text{O}$	0.00001	mg
$\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$	18	mg
NH_4Cl	15	mg
KH_2PO_4	1.6	mg
NaHCO_3	50	mg
$\text{MgCl}_2 \cdot 6\text{H}_2\text{O}$	12	mg
$\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$	15	mg

The constituents mentioned above were filled up to 1L with purified water.

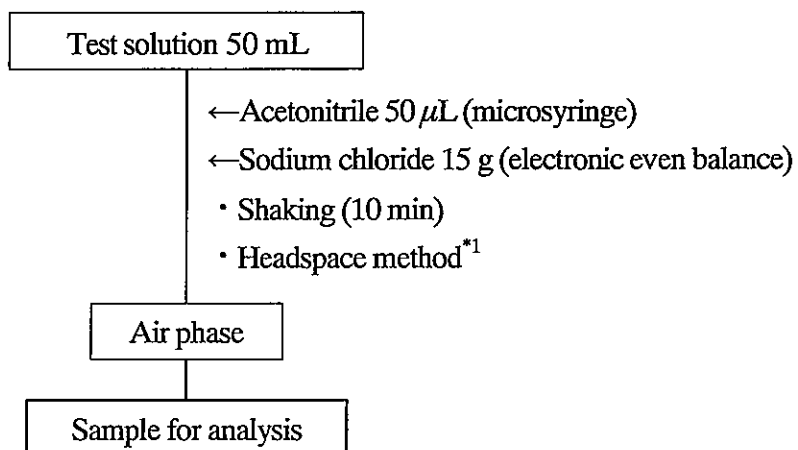
Appendix 2

Analytical method and measured concentration of test item

1. Pretreatment of test solution

The test solution sampled was pretreated according to the flow scheme to prepare the sample for analysis.

Flow scheme



*1 Headspace method condition

Vessel: 125 mL vial container

Warming: 70°C, more than 20 min

2. Method of analysis

The pretreated samples for analysis were quantitatively analyzed by gas chromatograph-mass spectrometry (GC-MS) under the following conditions to determine the concentration of the test item. The concentration of the test item in each sample for analysis was determined on the basis of a comparison of the peak area on the chromatogram of the sample with that of a standard sample. Some chromatograms obtained are shown in Appendix 3.

Analytical conditions

Instrument	Gas chromatograph-mass spectrometer
Gas chromatograph	Agilent 6890 Series Plus ⁺
Mass spectrometer	Agilent 5973N MSD

Gas chromatograph conditions

Column	HP-PONA film thickness 0.5 μ m (Agilent Technologies) 50 m \times 0.2 mm I.D. Fused silica
Column temperature	40°C(2 min) ^① \rightarrow 70°C(0 min) ^② \rightarrow 150°C(0.1 min)
Temp. rate	①15°C/min ②30°C/min
Carrier gas	Helium
Column flow	24.1 mL/min
Injection temp.	150°C
Injection volume	0.1 mL
Inlet mode	Split
Split ratio	13:1
Pressure	40 kPa

Mass spectrometer conditions

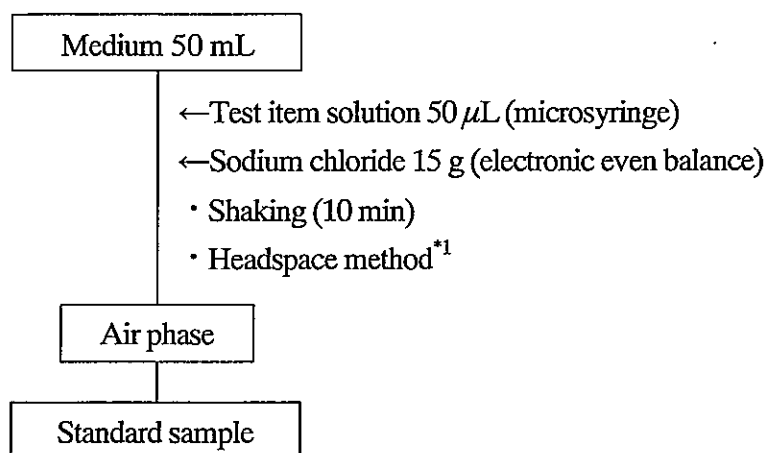
Ionization method	Electron ionization (EI)
Detecting method	Selected ion monitoring (SIM)
Measurement (m/z)	77
Ion source temp.	230°C
MS quadrupole temp.	150°C
Ionization voltage	69.9 eV
Interface temp.	200°C

3. Preparation of standard sample

The standard sample to determine the concentration of the test item in the sample for analysis was prepared as follows. The standard sample was prepared with correcting by the purity (99.8%) of the test item.

The test sample of 100.2 mg was precisely weighed with an electronic balance and dissolved in acetonitrile to obtain 1000 mg/L solution of the test item. The solution was diluted with acetonitrile to prepare 200 mg/L solution of the test item. And the solution was pretreated according to the flow scheme to prepare the 0.200 mg/L standard sample.

Flow scheme



4. Calibration curve

The test item solution of 20.0, 100, 200 and 400 mg/L were prepared by the same procedure as described in section 3. And they were pretreated according to the flow scheme of section 3 to prepare the standard sample of 0.0200, 0.100, 0.200 and 0.400 mg/L respectively. These samples were analyzed according to the quantitative analytical conditions described in section 2. A calibration curve was drawn from the relationship between the concentrations of standard sample and the peak area on the chromatogram, and the determination was confirmed. The calibration curve is shown in Appendix 3. The determination limit of the test item was the lowest concentration of the standard sample (0.0200 mg/L) within the range of the calibration confirmed. Therefore, the determination limit of the test item in the test solution was 0.0200 mg/L in consideration of pretreatment.

5. Results of measurement

The results of the measured concentrations of the test item in the test solution are shown below.

Appendix table 2-1 Measured concentrations of test item in test solutions

Nominal concentration (mg/L)	Measured concentration (mg/L)				
	At the start	24 hours	48 hours	At the end	Geometric mean
Control	n.d.	n.d.	n.d.	n.d.	
100	0.0712	n.d.* ²	n.d.* ²	n.d.* ³	0.0139

n.d. : < 0.0200 mg/L

*2 In accordance with OECD Guidance Document No.23, half the determination limit of the test item (0.0100 mg/L) was adopted for the calculation of geometric mean, because the test item was detected but not quantified.

*3 Because the test item was not detected, in accordance with OECD Guidance Document No.23, the detection limit (0.0147 mg/L) was attempt to adopt for the calculation of the geometric mean. But the value was higher than half the determination limit of the test item (0.0100 mg/L). Therefore 0.0100 mg/L was adopted for the calculation of geometric mean.

The geometric mean is calculated by the following expression:

$$\text{antilog} \left(\frac{1}{2(t_n - t_1)} \sum_{i=1}^{n-1} [(\log(\text{conc}_i) + \log(\text{conc}_{i+1})) \cdot (t_{i+1} - t_i)] \right)$$

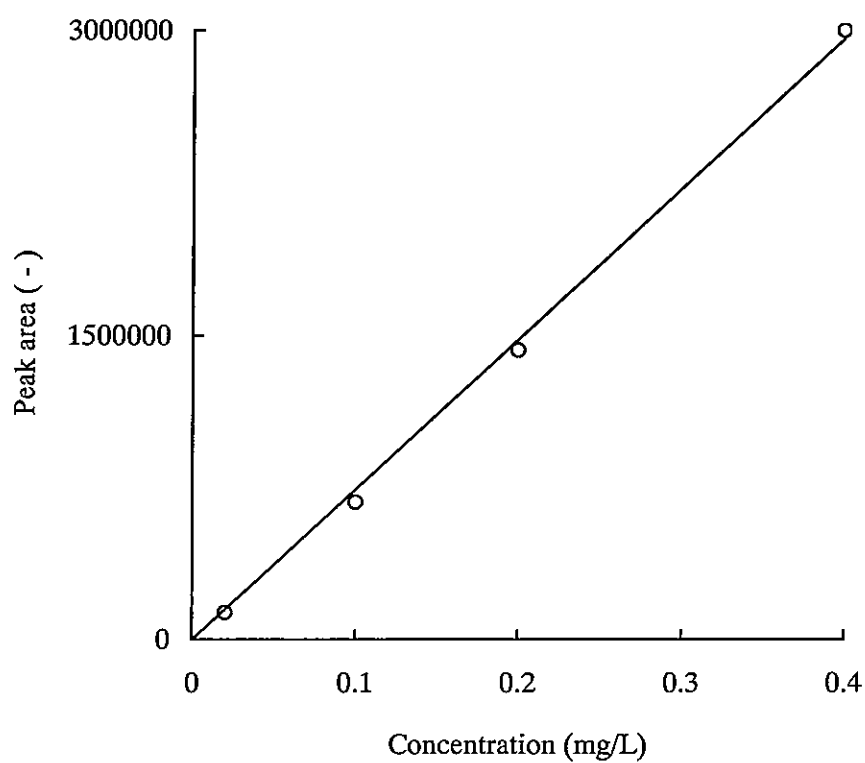
where

t_1 = at the start < t_2 < ... t_n = at the end

conc_1 = concentration at the start, $\text{conc}_2, \dots, \text{conc}_n$ = concentration at the end

Appendix 3

Calibration curve and chromatogram



$$y = 7395360x$$

$$r = 0.999$$

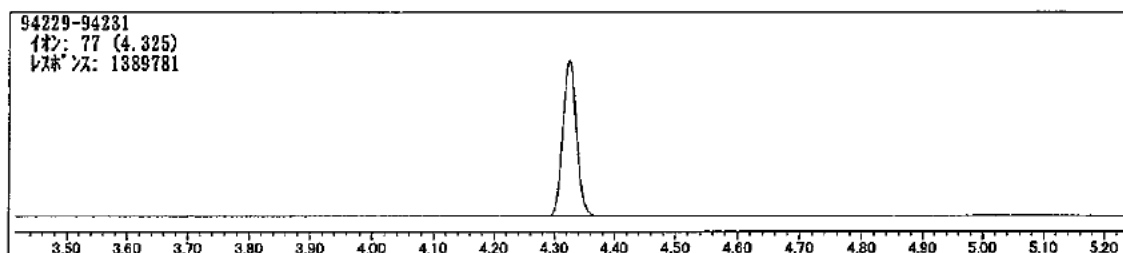
Concentration (mg/L)	Peak area (-)
0.0200	133392
0.100	678459
0.200	1428706
0.400	2999322

Appendix figure 3-1 Calibration curve of 13F-OLE for analysis by GC-MS.

Standard sample 0.200 mg/L

データファイル: C:\MSDCHEM\1\DATA\94229\070827\0827S02.D バイール: 1
 測定日: 27 Aug 2007 15:22 オペレータ:
 サンプル: Standard sample 0.200mg/L 装置: Instrument
 一般情報: 希釈率: 1.00
 サンプルアライメント: 0.00
 MS 積分パラメータ: autoint1.e
 定量メソッド: C:\MSDCHEM\1\METHODS\94231(B).M (ケミステーション インテグレーション)

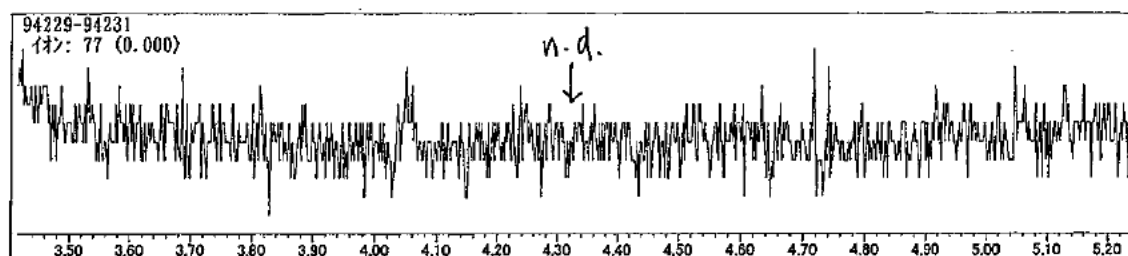
94229



Control

データファイル: C:\MSDCHEM\1\DATA\94229\070827\29H0H2.D バイール: 1
 測定日: 27 Aug 2007 15:34 オペレータ:
 サンプル: 94229 本試験 0h Control 装置: Instrument
 一般情報: 希釈率: 1.00
 サンプルアライメント: 0.00
 MS 積分パラメータ: autoint1.e
 定量メソッド: C:\MSDCHEM\1\METHODS\94231(B).M (ケミステーション インテグレーション)

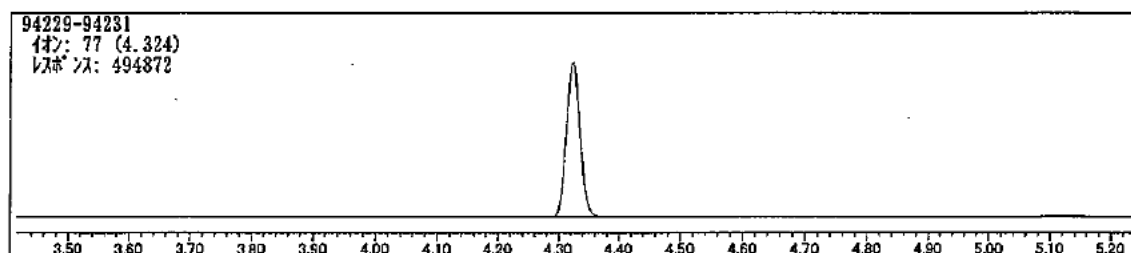
94229



100 mg/L (Nominal concentration)

データファイル: C:\MSDCHEM\1\DATA\94229\070827\290HA.D バイール: 1
 測定日: 27 Aug 2007 15:51 オペレータ:
 サンプル: 94229 本試験 0h 100mg/L 装置: Instrument
 一般情報: 希釈率: 1.00
 サンプルアライメント: 0.00
 MS 積分パラメータ: autoint1.e
 定量メソッド: C:\MSDCHEM\1\METHODS\94231(B).M (ケミステーション インテグレーション)

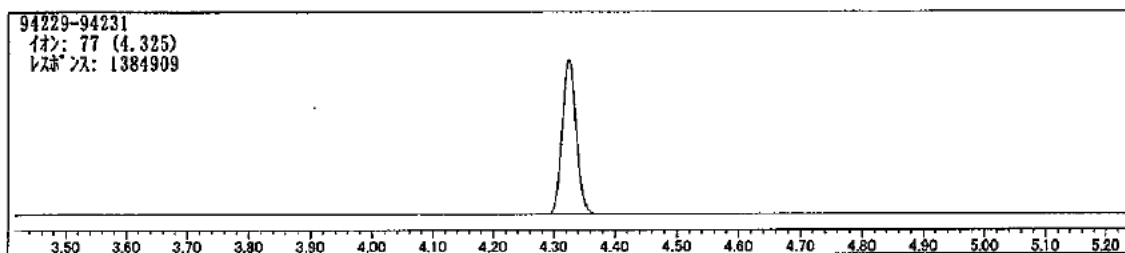
94229



Appendix figure 3-2 GC-MS chromatograms at start of exposure.

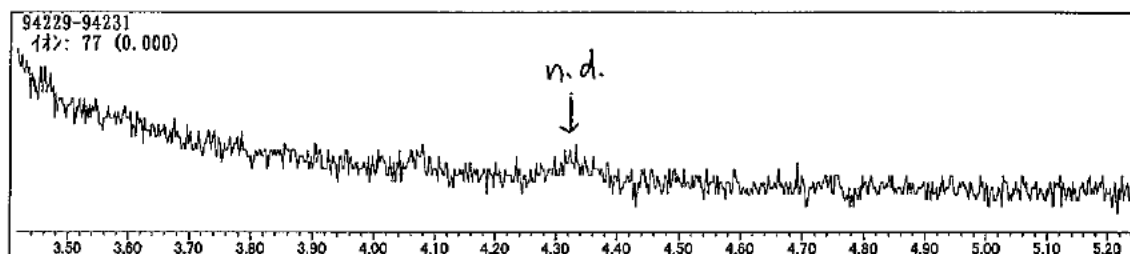
Standard sample 0.200 mg/L

データファイル : C:\MSDCHEM\1\DATA\94229\070830\0830SD1.D バイアス : 1
 測定日 : 30 Aug 2007 14:39 オペレータ :
 サンプル : Standard solution 0.200mg/L 装置 : Instrumen
 一般情報 : 希釈率 : 1.00
 サンプルアmount : 0.00
 MS 積分パラメータ : autoint1.e
 定量メソッド : C:\MSDCHEM\1\METHODS\94231 (B).M (ケミステーション インテグレーション) 94229



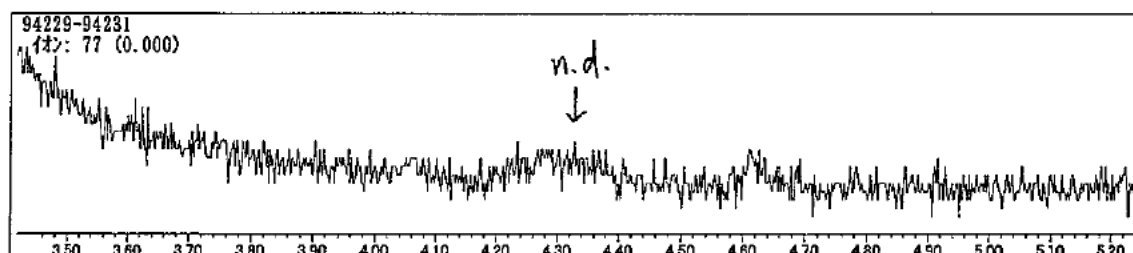
Control

データファイル : C:\MSDCHEM\1\DATA\94229\070830\29H72HZ.D バイアス : 1
 測定日 : 30 Aug 2007 14:51 オペレータ :
 サンプル : 93229 本試験 72h Control 装置 : Instrumen
 一般情報 : 希釈率 : 1.00
 サンプルアmount : 0.00
 MS 積分パラメータ : autoint1.e
 定量メソッド : C:\MSDCHEM\1\METHODS\94231 (B).M (ケミステーション インテグレーション) 94229



100 mg/L (Nominal concentration)

データファイル : C:\MSDCHEM\1\DATA\94229\070830\29H72HA.D バイアス : 1
 測定日 : 30 Aug 2007 15:02 オペレータ :
 サンプル : 93229 本試験 72h 100mg/L 装置 : Instrumen
 一般情報 : 希釈率 : 1.00
 サンプルアmount : 0.00
 MS 積分パラメータ : autoint1.e
 定量メソッド : C:\MSDCHEM\1\METHODS\94231 (B).M (ケミステーション インテグレーション) 94229



Appendix figure 3-3 GC-MS chromatograms at end of exposure.

Appendix 4

Solubility of test item in medium

1. Title

Solubility of test item in medium

2. Objective

The objective of this study is to estimate the solubility of the test item to medium.

3. Outline

Test item mixed with medium was stirred for 24 or 48 hours under the test temperature. After leaving at rest, the middle layer was sampled and analyzed.

4. Performance of test

4.1 Test equipments and instruments

Water bath :	Plastic tank
	Warming/cooling unit (Type HCA250, Sato craft)
Mixing apparatus :	Magnetic stirrer
Vessel :	Devised glass container (Interior volume : Approximately 600 mL)

4.2 Test conditions

(1) Test temperature :	23±1°C
(2) Measurement :	Twice (24 and 48 hours after the mixture was stirred)
(3) Repitition :	24 hours n=3 (Sample-1, Sample-2 and Sample-3)
	48 hours n=3 (Sample-4, Sample-5 and Sample-6)

4.3 Test procedures

- (1) Test sample and medium were mixed in a devised glass container to prepare approximately 100 mg/L* solution and sealed without headspace.

* The additive amount (38.5 μ L) was calculated from the density of the test item (1.560 g/cm³).

- (2) The test solution was stirred slowly using a magnetic stirrer under the test temperature in a water bath.
- (3) After the solution was stirred for 24 hours or 48 hours, the flask was settled in a water bath for approximately 1 hour.
- (4) After settling, middle layer was sampled and analyzed.

4.4 Analysis of test solution

- (1) Pretreatment of test solution

The middle layer of test solution was collected carefully from sampling spout of devised vessel by syringe. The collected solution was pretreated according to the flow scheme described in Appendix 2 1. Pretreatment of test solution.

- (2) Method for analysis

See Appendix 2 2. Method of analysis.

4.5 Preparation of standard sample

See Appendix 2 3. Preparation of standard sample.

4.6 Calibration curve

See Appendix 2 4. Calibration curve.

5. Results

Measured solubility of the test item after 48 hours was higher than that of after 24 hours. Therefore, value of after 48 was adopted to the solubility in medium. The solubility of the test item to medium was 0.177 mg/L. The results of analyses are shown below.

Appendix table 4-1 Value measured after stirring for 24 hours

Sample name	Measured value (mg/L)	Arithmetic mean (mg/L)
Sample-1	0.157	0.142
Sample-2	0.147	
Sample-3	0.121	

Appendix table 4-2 Value measured after stirring for 48 hours

Sample name	Measured value (mg/L)	Arithmetic mean (mg/L)
Sample-4	0.151	0.177
Sample-5	0.157	
Sample-6	0.223	

Additional data

Results of preliminary studies

1. Solubility of test item in medium

It was expected that the solubility of the test item in medium was below 100 mg/L, therefore, the measurement of the solubility of the test item in medium was conducted. The following preliminary study 1 was performed in fish acute toxicity test (Study number: 94231).

1) Preliminary study 1 for measurement of solubility

(1) Method

Since the test item was expected to be volatile due to the chemical structure, the test item and the dilution water (dechlorinated tap water) were mixed and gently stirred in a devised glass container under closed system with no head space and test temperature ($24\pm1^{\circ}\text{C}$) for 24 and 48 hours. For removal of insoluble substance, the procedure of gentle stirring and taking out from the middle layer of the settled solution for 1 hour was employed since centrifugation made the concentration of the test item decreased. The concentration of the test item in the collected sample was analyzed with pretreatment by gas chromatography - mass spectrometry (GC-MS). For 48 hours stirring, a sample of approximately 10 mg/L as nominal concentration was additionally measured.

(2) Result

Nominal concentration (mg/L)	Measured concentration (mg/L)	
	24-hour stirring hours	48-hour stirring
Approx. 100 (Sample-1)	0.0934	-
Approx. 100 (Sample-2)	0.129	-
Approx. 100 (Sample-3)	-	0.0949
Approx. 100 (Sample-4)	-	0.135
Approx. 10 (Sample-5)	-	0.0791

The measured value was around 0.1 mg/L with a little variance.

2) Preliminary study 2 for measurement of solubility

(1) Method

Similarly in the preliminary study 1, the test item and the medium were mixed and gently stirred in a devised glass container under closed system with no head space and temperature of algae growth inhibition test ($23\pm 1^{\circ}\text{C}$) for 24 and 48 hours. For removal of insoluble substance, the procedure of gentle stirring and taking out from the middle layer of the settled solution for 1 hour was employed. The concentration of the test item in the collected sample was analyzed with pretreatment by GC-MS.

(2) Result

Nominal concentration (mg/L)	Measured concentration (mg/L)	
	24-hour stirring hours	48-hour stirring
Approx. 100 (Sample-1)	0.262	-
Approx. 100 (Sample-2)	0.272	-
Approx. 100 (Sample-3)	-	0.352
Approx. 100 (Sample-4)	-	0.193

The solubility of the test item in medium was around 0.2 to 0.3 mg/L.

3) Summary of preliminary study for measurement of solubility

From the results of the preliminary studies, the solubility of the test item in the medium was estimated at around 0.2 to 0.3 mg/L. Since centrifugation made the concentration of the test item decreased, the procedure of gentle stirring and taking out from the middle layer of the settled solution for 1 hour was employed for removal of insoluble substance. Since the measured concentrations were almost same even in the different nominal concentrations of ten times (approximately 10 and 100 mg/L), it was thought that insoluble substance could be removed.

From the results mentioned above, in definitive study, the devised glass container would be used for the preparation, and the procedure of gentle stirring and taking out from the middle layer of the settled solution for 1 hour was employed in order to remove insoluble substance.

2. Study for effect on test organism

1) Preliminary study 1

(1) Method

The test sample and medium were mixed to prepare approximately 100 mg/L (nominal concentration) as upper limit of test concentration. Then, they were stirred for approximately 48 hours. After the mixture was left for approximately 1 hour at rest, the solution of test item was collected from the middle layer for use as the test solution. And then the test organisms were exposed to the test solution to confirm the effect. It was expected that the test item was volatile, therefore, the test solution was prepared under the closed condition without headspace, and the closed test vessel for volatile substance was used. At the same time, the concentration of the test item in the test solution was measured at the start and end of the exposure. The measurement in order to confirm whether the test item was absorbed to algae or not was not conducted because the operation like centrifugation could not be used for volatilization of test item.

(2) Result

Nominal concentration (mg/L)	Growth inhibition rate (%)
	Growth rate (0-3d)
Approx. 100	1.69

Replicates: two replicates / test level

Measurement method: cell counting method

The growth inhibition rate was approximately 1.7% in the saturated solution of the test item.

Nominal concentration (mg/L)	Measured concentration (mg/L) (Percent of the measured concentration to that at the start %)	
	At the start of the exposure	At the end of the exposure (After 72 hours)
Approx. 100	0.154	0.00410* (2.26)

* Reference value (below the determination limit)

The concentration of the test item at the start of the exposure was close to solubility in medium. The measured concentration of the test item significantly decreased at the end of the exposure.

2) Preliminary study 1

(1) Method

The test solution was prepared in the same manner as preliminary study 1 and was exposed to the test organisms to confirm the effect. The closed test vessel for volatile substance was used. At the same time, the concentration of the test item in the test solution was measured at the start and end of the exposure.

(2) Result

Nominal concentration (%)	Growth inhibition rate (%)
	Growth rate (0-3d)
10.0	3.08
30.0	-1.45
100	0.680

Replicates: two replicates / test level

Measurement method: cell counting method

The growth inhibition rate was approximately 0.7% in approximately 100 mg/L (nominal concentration) test level. Therefore, it was expected that the test item has no adverse effect in this level.

Nominal concentration (%)	Measured concentration (mg/L) (Percent of the measured concentration to that at the start %)	
	At the start of the exposure	At the end of the exposure (After 72 hours)
10.0	0.0108	0.00223 (20.6)
100	0.167	0.00223 (1.34)

In the 100% test level, the concentration of the test item at the start of the exposure was close to solubility in medium. The measured concentration of the test item significantly decreased at the end of the exposure.

3) Summary of effect on test organisms (results of preliminary study)

The test sample and medium were mixed to prepare the maximum concentration on the applied test guidelines (approximately 100 mg/L), and then they were stirred for 48 hours under closed condition. The mixture was left at rest for 1 hour, and then the solution was collected from the middle layer. The test organisms were exposed to the saturated solution (nominal concentration: approximately 100 mg/L). As a result, no effect was found in the saturated solution. The measured concentration of the test item significantly decreased during the exposure because of the volatilization of the test item. The exchange of the test solution was impossible in the algae growth inhibition study, and appropriate headspace in the test vessel was necessary for the growth of algae. Therefore, it was difficult to maintain the concentration of the saturated solution as a test design.

3. Operation of the definitive study

1) Measurement of solubility of test item in medium

Based on the results of the preliminary study for the solubility, measurement of the solubility of the test item in medium was conducted as follows. The test sample was added to medium to prepare the nominal concentration of approximately 100 mg/L. Then, they were stirred for 24 hours or 48 hours at $23\pm1^{\circ}\text{C}$ (temperature of algae growth inhibition study) in the closed vessel with little headspace. Removal of the insoluble matter such as the centrifugation and filtration was not conducted. After stirring, the test solution was left at rest for 1 hour, and then the solution collected from the middle layer. As the test solution, the concentration of the test item and hydrolyzed product was measured.

2) Definitive study

Based on the results of the preliminary study, it was expected that the test item did not affect the test organisms at around the solubility of the test item in medium. Therefore, the definitive study was conducted with the saturated solution (nominal concentration: approximately 100 mg/L) which was prepared by stirring for approximately 48 hours and the control. The preparation method of the test solution was as follows. The test sample and medium were added to the closed vessel without headspace to prepare approximately 100 mg/L, and they were stirred for approximately 48 hours by magnetic stirrer. The mixture was left at rest for 1 hour, then, the solution was collected from the middle layer, and this was used for the test solution (saturated solution of the test item). The concentrations of the test item in the test solution were measured at the start of the exposure, 24 and 48 hours after the exposure, and the end of the exposure.